

Particle properties of waves
Part-2
Problems on Blackbody Radiation

1. The wavelength of radiation emitted by a body depends upon
- (a) The nature of its surface
 - (b) The area of its surface
 - (c) The temperature of its surface
 - (d) All of the above factors

Solution : (c) The temperature of its surface

According to Wien's displacement law $\lambda_{max} \propto \frac{1}{T}$

So the wavelength of radiation emitted by a body only depends upon the temperature of its surface.

2. If a blackbody is heated at high temperature, it seems to be
- (a) Blue
 - (b) white
 - (c) Red
 - (d) Black

Solution : (b) White

When heated, depending upon the temperature of the cavity, the wavelength of radiation emitted by the blackbody changes. At first the body seems to be red then to yellow and at high temperatures the blackbody appears to be white.

- 3.If a black wire of platinum is heated, then its colour first appear red, then yellow and finally white. It can be understood on the basis of
- (a) Wien's displacement law
 - (b) Prevost theory of heat exchange
 - (c) Newton's law of cooling
 - (d) None of the above

Solution : (a) Wien's displacement law

Since Wien's displacement law defines the relation between wavelength and temperature.

4. Calculate the Sun's temperature assuming that it can be modified as a perfect blackbody. Also calculate the wavelength of maximum emission. Average solar power = $3.85 \times 10^{26} \text{ W}$, radius of sun = $6.96 \times 10^8 \text{ m}$

Given $P = 3.85 \times 10^{26} \text{ W}$

$r = 6.96 \times 10^8 \text{ m}$

$T = ?$

According to Stefan's Boltzmann law

$$\frac{P}{A} = a \sigma T^4 \rightarrow (1)$$

For a perfect blackbody emissivity $a = 1$

$$\sigma = 5.67 \times 10^{-8} \text{ W/m}^2 \text{ K}^4$$

From (1) $T^4 = \frac{P}{A\sigma}$

$$T = \sqrt[4]{\frac{P}{4\pi r^2 \sigma}}$$

$$= \sqrt[4]{\left(\frac{3.85 \times 10^{26}}{4\pi \times 5.67 \times 10^{-8} \times (6.96 \times 10^8)^2} \right)}$$

$T \approx 5559 \text{ K}$

Wavelength corresponding to maximum emission

$$\lambda_{max} T = \text{Constant} = 2.898 \times 10^{-3}$$

$$\lambda_{max} = \frac{2.898 \times 10^{-3}}{5559} = 501 \text{ nm}$$

5. Estimate the surface temperature of a star if the radiation it emits has a maximum intensity at a wavelength of 446 nm. What is the intensity radiated by the star ?

Here $\lambda_{max} = 446 \text{ nm}$

$$\lambda_{max} T = 2.898 \times 10^{-3}$$

$$T = \frac{2.898 \times 10^{-3}}{446 \times 10^{-9}}$$

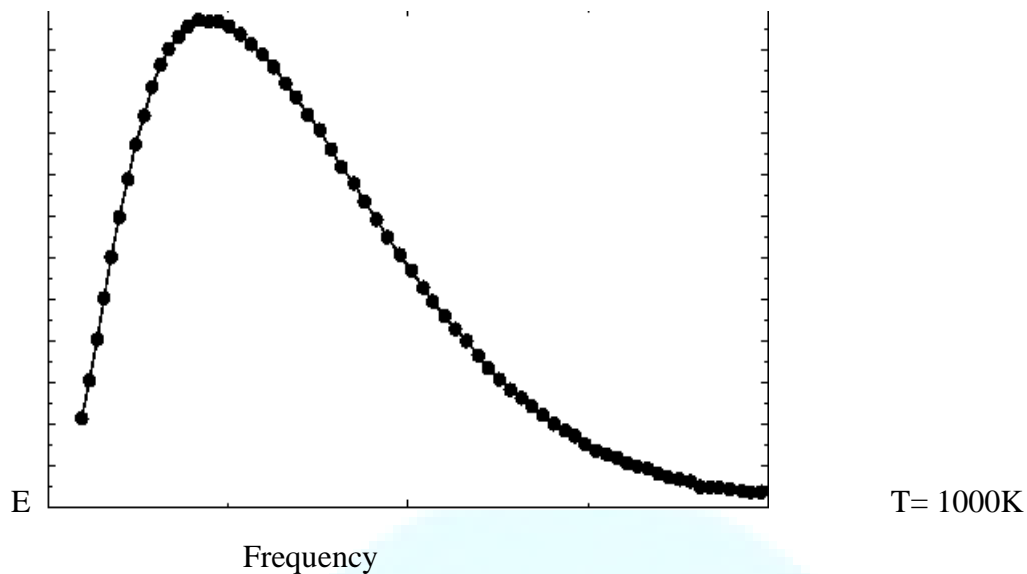
$T = 6500 \text{ K}$

Intensity, $I = a \sigma T^4$ ($a=1$)

$$I = 5.67 \times 10^{-8} \times 6500^4$$

$$= 101.2 \times 10^6 \frac{\text{W}}{\text{m}^2}$$

6. Calculate the frequency corresponding to the highest energy density from the curve given:



By Wien's displacement law

$$\lambda_{max} T = 2.898 \times 10^{-3} \rightarrow (A)$$

To find frequency we can use the relation between frequency and wavelength

$$C = \nu \lambda$$

$$\lambda = \frac{C}{\nu}$$

Substituting the value of λ in equation (A)

$$\frac{C}{\nu} T = 2.898 \times 10^{-3}$$

$$T = 1000 \text{ K}$$

$$\nu = \frac{CT}{2.898 \times 10^{-3}} = \frac{3 \times 10^8}{2.898 \times 10^{-3}} \times 1000$$

$$\nu \approx 1.03 \times 10^{14} \text{ Hz}$$

7. A maximum wavelength of $\lambda = 2 \mu\text{m}$ is emitted at 1600K. If the temperature is increased to 2000K, then the wavelength of the spectrum is —

We know, at a particular temperature $\lambda_{max} T = \text{constant}$

$$\lambda_1 = 2 \mu\text{m}$$

$$T_1 = 1600 \text{ K}$$

$$T_2 = 2000 \text{ K}$$

$$\lambda_2 = ?$$

$$\lambda_1 T_1 = \text{Constant} \text{ --- (1)}$$

$$\lambda_2 T_2 = \text{Constant} \text{ --- (2)}$$

From (1) and (2)

$$\lambda_1 T_1 = \lambda_2 T_2$$

$$\lambda_2 = \frac{2\mu\text{m} \times 1600}{2000} = 1.6 \mu\text{m}$$

8. The rate of radiation of a black body at 0°C is $E\text{J}/\text{sec}$. The rate of radiation of this black body at 273°C will be .

- (a) $16E$
- (b) $8E$
- (c) $4E$
- (d) E

Solution: (a) $16E$

By Stefan- Boltzmann law

$$E \propto T^4$$

$$T_1 = 0^\circ\text{C} = 273\text{K}$$

$$T_2 = 273^\circ\text{C} = 546\text{K}$$

$$E_1 = E$$

$$E_2 = ?$$

$$\frac{E_1}{E_2} = \frac{T_1^4}{T_2^4}$$

$$\frac{E_2}{E_1} = \frac{546^4}{273^4}$$

$$E_2 = 16E$$